

Exam. Code: 433  
Sub. Code: 2965

1115

M.Sc. (Applied Chemistry/Pharmaceutical)  
Third Semester

Paper – 303: Unit Pharmaceutical Operations

Max. Marks: 60

Time allowed: 3 Hours

**NOTE:** Attempt five questions in all, including Question No. I which is compulsory and selecting one question from each Unit.

x-x-x

I. Attempt the following:-

- a) What is mixing and agitation?
- b) Discuss the role of filter aid.
- c) Name industrial centrifuges.
- d) What is boiling point rise?
- e) Define diffusion coefficient and write its SI units.
- f) Give the classification of crystallizers. (6x2)

#### UNIT - I

- II. a) Discuss the mechanism of fluid mixing. (2x6)  
b) Write the construction and working of propeller mixer.
- III. a) Discuss different types of filtration processes.  
b) Give the classification of filters. And discuss the working of vacuum filter. (2x6)

#### UNIT - II

- IV. Explain the construction, working and industrial applications of basket centrifuge and sedimentation type centrifuge. Also discuss their advantages and disadvantages. (12)
- V. Discuss the following:
  - a) Flow properties of a powder or a bulk solid (2x6)
  - b) Compaction profiles

#### UNIT - III

- VI. a) What is evaporation? Give its industrial applications. State the difference between evaporation and distillation. (2x6)  
b) Draw a neat sketch of a rising film evaporator and label its parts.

P.T.O.



(2)

- VII. a) Write down the Fick's law of diffusion. Also discuss the applications of mass transfer operations at industrial scale.
- b) Oxygen is diffusing through carbon dioxide under steady-state conditions, with the carbon dioxide non-diffusing. The temperature and the total pressure of the system are respectively 273 K and  $101.3 \times 10^3 \text{ N/m}^2$ . The partial pressure of oxygen at two parallel, vertical planes 3 mm apart is respectively  $13000 \text{ N/m}^2$  and  $8000 \text{ N/m}^2$ . Determine the rate of diffusion of oxygen through per  $\text{m}^2$  of the two planes. Diffusivity of the gas mixture is  $1.39 \times 10^{-5} \text{ m}^2/\text{s}$ . (2x6)

#### UNIT- IV

- VIII. a) What is VLE, Dalton's law, Henry's law and relative humidity?
- b) Discuss the various steps to calculate number of plates by Mc-Cabe Thiele method. Also write various assumptions. (4,8)
- IX. a) Explain the mechanism of crystallization process. Also Name different types of crystallizers.
- b) Discuss the mechanism of drying and develop a relation to find out time of batch drying. (2x6)

x-x-x