Exam.Code: 0003 Sub. Code: 0277

#### 2122

## B.A./B.Sc. (General) Third Semester Industrial Chemistry Paper - A: Material Science

Time allowed: 3 Hours Max. Marks: 75

**NOTE**: Attempt <u>five</u> questions in all, including Question No. IX (Unit-V) which is compulsory and selecting one question each from Unit I - IV.

	x-x-x	
	<u>UNIT-I</u>	
I.	a) Explain in detail the manufacturing process of cement.	
	b) Write a complete note on Super Alloys.	(9,6)
II.	Write notes on the following:-	
*	a) Alloy steels	
	b) Mercerization	
	c) Moulding of plastics into articles	(3x5)
	<u>UNIT – II</u>	
III.	a) Discuss kinetics and mechanism of process of acetanilide to P-nitroacetanilide.	
	b) Explain kinetics of halogenations reactions.	(8,7)
IV.	a) Discuss in detail commercial manufacturing of chloromethanes.	
	b) Explain different reagents for halogenations.	(8,7)
	<u>UNIT – III</u>	

- V. Describe commercial manufacturing of the following:
  - a) Maleic anhydride
  - b) Cellulose acetate

(7,8)

- VI. Discuss mechanism of following:
  - a) Vapour phase oxidation
  - b) Hydrolysis

(8,7)

### UNIT-IV

- VII. Discuss in brief commercial manufacturing of following:
  - a) Ethyl benzene
  - b) Phenyl ethyl alcohol
  - c) P-aminophenol

(3x5)

# VIII. Explain the following:-

- a) Commercial manufacturing of Aniline
- b) Hydrogenation of vegetable oil

(7,8)

### UNIT - V

- IX. Answer the following in brief:
  - a) Spinning of fibers
  - b) High speed steels
  - c) Commercial manufacturing of Benzoic acid
  - d) Hydrolyzing agents
  - e) Metal and alkali reductions

(5x3)

x-x-x