(i) Printed Pages: 3

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(ii) Questions :9

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B.A./B.Sc. (General) 6th Semester (2053)

CHEMISTRY

(Same for B.Sc. Microbial and Food Technology)

Paper: XXII: Organic Chemistry-B

Time Allowed: Three Hours]

[Maximum Marks: 22

Note: Attempt any five questions in all including question No. 9 which is compulsory question and selecting one question from each Unit I-IV.

UNIT-I

- (a) What are α-amino acids? Give evidence supporting dipolar ionic structures of these acids.
 - (b) Give the concept of electrophoresis. What is its importance in protein chemistry?
- 2. (a) Give a brief description of the solid-phase peptide synthesis.
 - (b) Write down the structure and importance of Sanger's reagent in protein chemistry. 2,2

UNIT-II

- (a) Discuss free radical vinyl polymerization.
 - (b) Explain the mechanism of anionic polymerization. Why does styrene undergo anionic polymerization? 2,2
- (a) What is condensation polymerization? Discuss the preparation of Nylon-66.
 - (b) How is Bakelite formed? Explain the reactions with equations. 2,2

UNIT-III

- (a) Acetoacetic ester is an equilibrium mixture of keto and enol forms. Give evidence in support of the statement.
 - (b) How can you convert cyclohexanone into 2-hexanoyl cyclohexanone through the intermediate formation of an enamine?
 2,2
- 6. (a) What are enolates? Give two examples of their synthetic utility.
 - (b) Write all resonating structures of carbanion of malonic ester.

2,2

UNIT-IV

- 7. (a) What is the role of dry ether in the preparation of Grignard's reagents?
 - (b) Give the reaction of Grignard's reagents with α, β-unsaturated carbonyl compounds.
 2,2

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- 8. (a) Give one method of preparation of Grignard's reagent and diethyl zinc.
 - (b) Write the chemical equation for preparation of ethylmagnesium bromide. 2,2

(Compulsory Question)

- 9. (a) Define acid-base behaviour of amino acids.
 - (b) What are epoxy resins?
 - (c) Define the term keto-enoltautomerism.
 - (d) Define the term organolithium compounds. $4 \times 1.5 = 6$