### 2123

# B.A./B.Sc. (General) Third Semester **Industrial Chemistry** Paper - A: Material Science

Time allowed: 3 Hours

Max. Marks: 75

NOTE: Attempt five questions in all, including Question No. I which is compulsory and selecting one question from each Unit.

- I. Write brief note on following:
  - a) Write down the applications of alloys.
  - b) Define ultimate tensile strength.
  - c) Give examples of commonly used oxidizing agents for organic compounds.
  - d) Write down any Friedel-Craft alkylation reaction.
  - e) Define esterification.
  - f) Write down the product obtained upon the nitration of nitrobenzene.
  - g) Define strain.
  - h) What is batch nitration?
  - Give examples of reagents used for halogenation reactions.
  - j) Give example of a metal-induced reduction reaction.  $(10 \times 1.5)$

## UNIT-I

- a) Define ceramics. Write down their applications. Explain their manufacturing II. process in detail.
  - b) Differentiate between metals and alloys.

(12, 3)

- a) Explain the manufacturing process of cement in detail.
  - b) Write a note on the composition of different types of cement.
  - c) Explain the science behind the setting of cement.

(8, 4, 3)

#### UNIT-II

- a) Write in detail about the commercial manufacturing of chloromethane.
  - b) Differentiate between aromatic side chain and nuclear halogenations.

(10, 5)

- a) What is nitration? Discuss about various nitrating agents.
- b) Give the mechanism of nitration of acetanilide to give p-nitroacetanilide. (5, 10)

#### UNIT-III

- a) Discuss in detail the commercial manufacturing of maleic anhydride.
  - b) Expalin liquid-phase oxidation in detail.

(10, 5)

- VII. a) Write a note on the manufacturing of cellulose acetate.
- b) Discuss the mechanism of: (i) Acid-hydrolysis of esters, (ii) Basic-hydrolysis of (9, 6)esters.

## UNIT-IV

- VIII. a) Write in detail about the commercial manufacturing of p-aminophenol.
  - b) Explain the kinetics of hydrogenation reaction.

(10, 5)

- a) Write a note on the manufacturing of ethylbenzene. IX.
  - b) Explain the thermodynamics of a hydrogenation reaction.

(10, 5)