(1)	Printed Pag	es: 3	Roll No					
(ii)	Questions	: 9	Sub. Code:	0	1	5	2	
			Exam. Code:	0	0	0	2	

B.A./B.Sc. (General) 2nd Semester (2054)

CHEMISTRY

Paper-VI: Organic Chemistry-B (Same for B.Sc. Microbial & Food Tech.)

Time Allowed: Three Hours] [Maximum Marks: 22

Note:—Attempt FIVE questions in all including Q. No. 9 which is compulsory question and selecting ONE question each from Units I to IV.

UNIT-I

- (a) Explain the mechanism of dehydrohalogenation of 2-bromobutane.
 - (b) Predict the products of the following reaction :

$$CH_3CH_2CH=CH_2 \xrightarrow{O_3} ? \xrightarrow{2[H]} ? \longrightarrow 3,1$$

 (a) Write the product/s and depict the steps involved in the mechanism of following reaction:

$$CH_3CH=CH_2 \xrightarrow{(BH_3)_2} ?$$

(b) What happens when ethylene is treated with Br_2 and H_2O ?

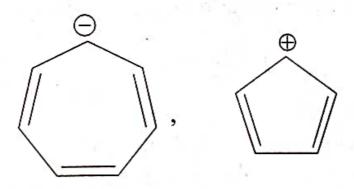
UNIT-II

- 3. (a) What will be the product of the reaction of cyclohexene with NBS?
 - (b) Elaborate the effect of temperature on 1,2 and 1,4-additions of HBr to 1,3-butadiene. 1,3
- 4. (a) Describe the mechanism of nucleophilic addition reaction of alkynes.
 - (b) With chemical equation, predict the product/s of reduction of 2-buytne with following:
 - (i) Na/NH₃
 - (ii) Pd/BaSO₄.

2,2

UNIT—III

5. (a) Which of the following compounds are aromatic? Explain with Huckel rule.



(b) Describe the role of σ and π - complexes in aromatic electrophilic substitution. 2,2

- 6. (a) Why chlorination of chlorobenzene is more difficult than that of toluene?
 - (b) Depict the mechanism of sulphonation of benzene.

2,2

UNIT-IV

- 7. (a) With mechanism, explain why reaction of 1-iodo-2, 2-dimethylpropane with water provide 2-methylbutan-2-ol instead of 2,2-dimethylpropan-1-ol?
 - (b) Compare the relative reactivity of benzyl chloride and vinyl chloride in nucleophilic substitution reactions.

2,2

- 8. (a) Elaborate the addition-elimination mechanism of nucleophilic aromatic substitution reactions.
 - (b) Depict the conversion of benzyl chloride to toluene.

3.1

(Compulsory Question)

- 9. (a) Out of BHr and HCl which gives peroxide effect and why?
 - (b) Explain Diels-Alder Reaction.
 - (c) What happens when toluene is treated with chromyl chloride?
 - (d) How will you prepare carbon tetrachloride?

 $4 \times 1.5 = 6$