(i) Printed Pages: 3 Roll No.

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B.A./B.Sc. (General) 6th Semester (2054)

CHEMISTRY

(Same for B.Sc. Microbial and Food Technology)

Paper-XXI: Inorganic Chemistry-B

Time Allowed: Three Hours [Maximum Marks: 22

Note:—Attempt FIVE questions in all, selecting ONE question each from Units I to IV. Unit V is compulsory.

UNIT-I

- 1. (a) What are Silicon Rubbers? How are these vulcanized?
 - (b) How does the delocalization of π -system in cyclic triphosphazene differ from π -system in benzene?

2,2

- 2. (a) What are Silicones fluids or oils? Discuss their uses.
 - (b) How is cyclic (NPCl₂)₃ prepared? Discuss the structure and bonding in (NPCl₂)₃.

UNIT-II

- (a) What is Symbiosis? Discuss theoretical basis of hardness and softness.
 - (b) How does HSAB principle explain the validity of the following reactions?

$$CHF_3 + CH_3F \rightarrow CF_4 + CH_4$$

 $CaS + H_2O \rightarrow CaO + H_2S$

2,2

- 4. (a) What is HSAB principle.? How does HSAB principle govern the occurrence of minerals?
 - (b) Explain the relationship between electronegativity and hardness.2,2

UNIT—III

- (a) Calculate Ground state terms symbol and number of microstates for d² and d³ octahedral systems.
 - (b) Discuss briefly L-S Coupling.

2,2

- 6. (a) Draw and discuss the Orgel diagram for $[Cu(H_2O)_6]^{+2}$ ion.
 - (b) Explain Laporte forbidden transitions.

2,2

UNIT—IV

- 7. (a) Give two main disadvantages of Gouy's method used for measuring the magnetic susceptibility?
 - (b) Explain the following:-
 - (i) TIP
 - (ii) Diamagnetic Correction
 - (iii) Anti-Ferromagnetism.

1,3

- 8. (a) Discuss variations of magnetic susceptibility with temperature.
 - (b) Explain Orbital contribution to magnetic moment in complexes. 2,2

UNIT-V

- (a) Draw the general repeating unit in silicone and phospazene.
 - (b) Discuss the applications of HSAB Principle.
 - (c) Calculate the spin only magnetic moments for Fe⁺² and Ni⁺² ions.
 - (d) $[Ti(H_2O)_6]^{+3}$ is purple in colour, Explain. 1.5×4