Exam. Code: 1368 Sub. Code: 9401

2054

B. Voc. (Medical Lab Technology) Second Semester

BMLT-204: Basics of Biochemistry

Time allowed: 3 Hours

Max. Marks: 40

NOTE: Attempt five questions in all, including Question No. I which is compulsory and selecting one question from each Unit.

X-X-X

- I. Answer the following:
 - a) Define acid.
 - b) What is the difference between molarity and molality?
 - c) Give the name one base?
 - d) Define colloids.
 - e) What do you mean by surface tension?
 - f) What is the flame photometer?
 - g) Difference between spectrophotometer and colorimeter.
 - h) Give two names of aromatic compounds.

(8x1)

UNIT - I

- II. a) Briefly describe the buffer solution and its significance.
 - b) What is the pH? Write one method by which pH is measured.

(2x4)

- III. a) Write about the applications of radioactive materials in biochemistry.
 - b) Describe in brief about normal and molar solution.

(2x4)

UNIT - II

- a) Briefly describe the characteristics of organic compounds.
 - b) What is the dialysis? Write its importance in Biochemistry.

(2x4)

- V. a) Write about classification of colloids.
 - b) Differentiate between organic and inorganic compounds.

(2x4)

UNIT - III

- VI. a) Give the names of major classes of centrifugations and their applications.
 - b) Write the significance of electrophoresis.

(2x4)

- VII. a) Explain about the spectrophotometers. Also write about its applications in labs.
 - b) Briefly describe the urinometer and its application in urine analysis.

(2x4)

<u>UNIT - IV</u>

- VIII. a) Write the major characteristics of proteins.
 - b) Write the basic characteristics of nucleic acid, and classified them.

(2x4)

- IX. a) What is the carbohydrate metabolism? Write only the names of related metabolic pathways.
 - b) Briefly describe about enzyme. Why enzymes are important in biological system? (2x4)