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(i) Printed Pages: 3 Roll No.

(ii) Questions :9 Sub. Code: 1 7 2 5 1 Exam. Code: 0 0 0 3

B.A./B.Sc. (General) 3rd Semester (2124)

CHEMISTRY

Paper—X: (Organic Chemistry-A)
(Same for B.Sc. Microbial & Food Technology)

Time Allowed: Three Hours] [Maximum Marks: 22

Note: — There are total NINE questions. Attempt FIVE questions in all selecting at least ONE question from each unit. Question No. 9 is compulsory.

UNIT—I

- (a) Arrange the alcohols in order of reactivity towards HBr, and justify your answer:
 2-butanol, 2-methyl-1-propanol, 2-methyl-2-propanol.
 - (b) Write the Chemical equation for reduction of butanoic acid with LiAIH₄.
- 2. (a) Write the mechanism of Claisen rearrangement. 2
 - (b) Compare the acidic character of Phenol and Ethanol and explain your answer.2

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UNIT-II

- (a) Butane nitrile is treated with CH₃MgBr, and then hydrolysed.
 Write the chemical equations and predict the product formed.
 - (b) Ethanal and Propane, both have same molecular mass i.e. 44, yet ethanal has higher B.Pt, why?
- 4. (a) Write the possible structures of two different carbonyl compounds having molecular formula C₈H₈O that contain a benzene ring.
 - (b) How will you synthesize butanal from butanoyl chloride?
 Write the chemical equation mentioning necessary inorganic reagents required.

UNIT-III

5. (a) What starting materials are required to synthesize the following alkene using Witting reaction?

2

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- (b) A Mannich reaction is carried out using cyclohexanone, formaldehyde and Diethyl amine. Write the chemical equation and the product formed.
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- 6. (a) Which of the following will undergo Cannizarro reaction and why: Acetaldehyde, formaldehyde, benzaldehyde and benzophenone?
 - (b) Write the mechanism of Knoevenagel reaction.

UNIT-IV

7. (a) Complete the reaction:

$$OH + C_2H_5OH \xrightarrow{H_2SO_4} ?$$

- (b) $CH_3CH_2COOH + Br_2/P_4 \xrightarrow{\Delta}$?
- (c) $C_6H_5COOH + NH_3 \longrightarrow ?$
- (d) $ClCH_2COONa + KCN/H_3O^+ \longrightarrow ?$ 1×4
- (a) Arrange the following in increasing order of their acidity:
 Methanoic acid, Acetic Acid, Chloroacetic acid, Fluoroacetic acid. Explain your answer.
 - (b) Benzyl Chloride is treated with KCN and product A is formed, A is then hydrolysed under acidic conditions. Identify A and B.

(Compulsory Question)

- 9. (a) Which is more acidic out of water and ethanol and why?
 - (b) What happens when propane 1, 3-dioic acid is strongly heated?
 - (c) Write chemical equation for reaction between Formadlehyde and hydroxyl amine under appropriate conditions.
 - (d) What reagents are needed in Baeyer Villiger oxidation of ketones?
 - (e) Write the equation for Clemmensen reduction of Butan-2-one.
 - (f) Draw the structure and write name of citric acid.

 $1 \times 6 = 6$