Exam.Code:0002 Sub. Code: 0152

#### 2071

# B.A./B.Sc. (General) Second Semester Chemistry

Paper – VI: Organic Chemistry – B

(Same for B.Sc. Microbial and Food Technology)

Time allowed: 3 Hours

Max. Marks: 22

**NOTE:** Attempt <u>five</u> questions in all, including Question No. IX (Unit-V) which is compulsory and selecting one question each from Unit I-IV.

x-x-x

### UNIT - I

- I. Describe the followings in relevance to alkenes:
  - a) Oxymercuration-reduction
  - b) Regioselectivity in alcohol dehydration.

(2,2)

II. Predict the products of following reactions:-

(i) 
$$CH_3CH_2CH=CH_2 \xrightarrow{O_3} ? \xrightarrow{2[H]} ?$$
  
(ii)  $? \xrightarrow{Hot \ KMNO_4/KOH} CH_3CH=CH_2 \xrightarrow{773 \ K} ?$ 

(4)

### UNIT - II

- III. a) Discuss the mechanism of nucleophilic addition reaction of alkynes.
  - b) Explain the acidity of alkynes.

(3,1)

- IV. a) Depict the 1,2 additions to conjugated dienes with appropriate example.
  - b) Elaborate metal-ammonia reduction of acetylene.

(2,2)

## **UNIT - III**

- V. a) Differentiate between aromatic and non-aromatic compounds.
  - b) Describe the mechanism of nitration of toluene.

(1,3)

P.T.O.

- VI. Discuss the following in relation to aromatic electrophilic substitution:
  - a) Role of  $\sigma$  and  $\pi$  -complexes
  - b) Orientation and ortho/para ratio

(2,2)

### **UNIT-IV**

- VII. a) Elaborate the elimination-addition mechanism of nucleophilic substitution in aryl halide.
  - b) How will you prepare dibenzyl from benzyl chloride?

(3,1)

- VIII. a) Discuss the mechanism of  $S_N^2$  reaction with energy profile diagram.
  - b) Predict the product of the following reaction with appropriate reasoning.

$$\frac{\text{CH}_2\text{-CH}_3}{\frac{\text{Br}_2}{h\nu}}?$$

(2,2)

#### UNIT - V

- IX. Attempt the following:
  - a) Illustrate Saytzeffs rule with example.
  - b) Explain the Diels-Alder reaction with suitable example.
  - c) How aromaticity is related to Huckel rule?
  - d) Give the preparation of carbon tetrachloride from carbon disulphide.