Exam.Code:0432 Sub. Code: 3449

2071

M.Sc. (Applied Chemistry/Pharmaceutical) Second Semester Paper – 203: Analytical Chemistry

Time allowed: 3 Hours

Max. Marks: 60

NOTE: Attempt <u>five</u> questions in all, including Question No. I which is compulsory and selecting one question from each Unit.

x-x-x

- I. Attempt the following:
 - a) What is HPTLC? Tell briefly.
 - b) How IR spectroscopy differs from Raman spectroscopy? Write differences in tabular form.
 - c) Write a brief note on silver-silver chloride electrode.
 - d) What are applications of refractometry?

(4x3)

UNIT-I

- II. a) Explain principle and construction of combination glass electrode with applications.
 - b) What is Ostward dilution law? Derive an expression for it.

(6,6)

- III. a) Discuss in detail about amperometric titrations. Tell advantages and disadvantages also.
 - b) Explain theory, working and applications of polarography.

(6,6)

<u>UNIT – II</u>

- IV. a) Discuss principle and working of refractometer with schematic diagram.
 - b) Discuss and compose CD and ORD. (Circular Dichroism and Optical Rotatory Disperssion) (6,6)
- V. a) Discuss basic principle of atomic absorption spectrophotometry. How it differs from flame emission spectrophotometry?
 - b) Write in detail about continuous and discrete atomizers.

(6,6)

<u>UNIT – III</u>

- VI. a) Discuss in detail Craige's method of multiple extraction.
 - b) Write a complete note on continuous counter-current extraction.

(6,6)

VII. a) Discuss in detail theory, principle and applications of Raman Spectroscopy.

b) How liquid-liquid extraction differs from liquid-solid extraction. Tell with examples. (6,6)

<u>UNIT - IV</u>

VIII. Discuss the following in detail:-

a) Dupont Analyzer

b) Electrolytic hygrometer

(6,6)

IX. a) Discuss in detail about GLC.

b) Write a note on Super Critical Fluid Chromatography.

(6,6)