Exam.Code: 0001 Sub. Code: 0083

2012

B.A./B.Sc. (General) First Semester Industrial Chemistry

Paper - A: Industrial Aspect of Chemistry

Time allowed: 3 Hours

Max. Marks: 75

NOTE: Attempt <u>five</u> questions in all, including Question No. I which is compulsory and selecting one question from each Unit.

X-X-X

- I. Answer the following:
 - a) What are silicates?
 - b) What is cellulose acetate?
 - c) Define homogeneous and heterogeneous catalysis.
 - d) Define gels.
 - e) Explain Roasting.
 - f) Write molecular formula and structure of furfural.
 - g) Define adsorption Isotherm.
 - h) Write advantages of coal.
 - i) Define and explain surfactants.
 - i) What is dextrin?

 $(10x1\frac{1}{2})$

UNIT - I

- II. Explain following chemical processes:
 - a) Fractionation of crude oil
 - b) Isomerisation
 - c) Hydroforming

(3x5)

- III. Write a note about:
 - a) Properties and structure of coal
 - b) Reforming
 - c) Natural gas

(3x5)

UNIT - II

IV. Explain manufacturing process of viscose and dextrose.

(15)

- V. Write a note on following metallurgical operations:
 - a) Calcination
 - b) Refining
 - c) Pulverisation

(3x5)

P.T.O.

<u>UNIT – III</u>

VI. Discuss physiochemical extraction of aluminium and lead. (15)

VII. Write a detailed note on various inorganic materials of industrial importance discussing their availability, forms and structure. (15)

$\underline{UNIT} - IV$

VIII. Discuss in detail about the following:-

- a) Phase transfer catalysis
- b) Sols
- c) Aerosols (3x5)

IX. Write in detail about:-

- a) Emulsions
- b) Effect of surfactants

c) Negative catalysis (3x5)

x-x-x